Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$)

$U_{\rm eq} = (1/3) \sum_i \sum_j U^{ij} a_i^* a_i^* \mathbf{a}_i \cdot \mathbf{a}_i.$

	x	v	z	U_{eo}
Bal	0.383679 (14)	0.000292 (11)	0.214799 (8)	0.01764 (4)
S1	0.12012 (8)	0.33320 (6)	0.35348 (4)	0.03023 (11)
01	0.5357 (2)	0.1509 (2)	0.27214 (14)	0.0351 (4)
O2	-0.0002(3)	0.4130 (2)	0.2886 (2)	0.0439 (5)
O3	0.0499 (3)	0.2870 (2)	0.4441 (2)	0.0482 (5)
N1	0.2534 (3)	0.2194 (2)	0.3100 (2)	0.0308 (4)
C1	0.2670 (3)	0.4096 (2)	0.3645 (2)	0.0289 (4)
C2	0.2472 (4)	0.5155 (3)	0.3999 (2)	0.0424 (6)
C3	0.3920 (5)	0.5510 (3)	0.3982 (2)	0.0498 (8)
C4	0.5481 (5)	0.4832 (3)	0.3649 (2)	0.0462 (7)
C5	0.5665 (4)	0.3758 (2)	0.3314 (2)	0.0337 (5)
C6	0.4225 (3)	0.3407 (2)	0.33174 (14)	0.0256 (4)
C7	0.4106 (3)	0.2286 (2)	0.3013 (2)	0.0264 (4)
S2	0.54351 (6)	0.17306 (5)	0.02100 (3)	0.01965 (8)
04	0.6980 (3)	0.2292 (2)	-0.22147(11)	0.0329 (4)
O5	0.6417 (2)	0.06171 (15)	0.08081 (10)	0.0241 (3)
O6	0.3676 (2)	0.2105 (2)	0.05516 (12)	0.0272(3)
N2	0.5690 (3)	0.1554 (2)	-0.08385 (12)	0.0250 (3)
C11	0.6325 (3)	0.2978 (2)	0.00359 (14)	0.0205 (3)
C12	0.6501 (3)	0.3649 (2)	0.0656 (2)	0.0251 (4)
C13	0.7304 (3)	0.4578 (2)	0.0306 (2)	0.0307 (4)
C14	0.7899 (3)	0.4805 (2)	-0.0620(2)	0.0319 (5)
C15	0.7707 (3)	0.4117 (2)	-0.1231(2)	0.0273 (4)
C16	0.6908 (3)	0.3193 (2)	-0.08868(14)	0.0217 (3)
C17	0.6546 (3)	0.2325 (2)	-0.13926(14)	0.0240 (4)
07	0.6926 (2)	-0.1701 (2)	0.27442 (12)	0.0293 (3)
08	0.4163 (2)	-0.1405 (2)	0.39619 (11)	0.0273(3)
09	0.0946 (2)	-0.0056 (2)	0.33593 (11)	0.0256 (3)
010	0.0781 (2)	0.0620 (2)	0.14379 (12)	0.0313 (3)
C21	0.8110 (4)	-0.2443 (3)	0.2153 (2)	0.0435 (6)
C22	0.7099 (3)	-0.2252 (2)	0.3698 (2)	0.0309 (4)
C23	0.5724 (3)	-0.1454 (3)	0.4255 (2)	0.0304(4)
C24	0.2729 (3)	-0.0999(3)	0.4583 (2)	0.0322 (5)
C25	0.1216 (3)	-0.1018 (2)	0.4171 (2)	0.0301 (4)
C26	-0.0493 (3)	0.0011 (3)	0.2927 (2)	0.0357 (5)
C27	-0.0668 (3)	0.1009 (3)	0.2060 (2)	0.0353 (5)
C28	0.0520 (3)	0.1254 (3)	0.0501 (2)	0.0341 (5)

Table 2. Selected geometric parameters (Å, °)

Ba1-O10	2.741 (2)	S1C1	1.772 (3)	
Ba1O1	2.787 (2)	01—C7	1.239 (3)	
Bal—O8	2.807 (2)	NI	1.343 (3)	
Ba1-09	2.814 (2)	S206	1.450 (2)	
Ba1—O7	2.823 (2)	S2—O5	1.453 (2)	
Ba1—O5	2.867 (2)	S2N2	1.598 (2)	
Bal-O6	2.945 (2)	S2C11	1.766 (2)	
Bal—N1	2.995 (2)	O4C17	1.232 (3)	
S1—O2	1.441 (2)	O4Bal'	2.898 (2)	
S1—O3	1.442 (2)	N2-C17	1.365 (3)	
S1—N1	1.620 (2)	N2—Bal'	2.848 (2)	
O2—S1—O3	115.73 (14)	O6—S2—O5	112.08 (10)	
02—S1—NI	111.41 (13)	O6-S2-N2	112.80 (11)	
O3—S1—N1	111.21 (13)	O5	111.14 (10)	
02—\$1—C1	110.59 (12)	O6-S2-C11	111.52 (10)	
03-S1-C1	109.48 (13)	O5-S2-C11	110.98 (10)	
N1—S1—C1	96.81 (11)	N2-S2-C11	97.49 (10)	
C7—N1—S1	111.3 (2)	C17—N2—S2	111.61 (15)	
C6-C1S1	106.8 (2)	C16-C11-S2	106.86 (15)	
C2C1S1	131.4 (2)	C12-C11-S2	130.3 (2)	
01—C7—N1	123.7 (2)	O4-C17-N2	122.9 (2)	
O1C7C6	122.4 (2)	O4-C17-C16	124.4 (2)	
N1C7C6	113.9 (2)	N2C17C16	112.7 (2)	
Symmetry code: (i) $1 - r - v - z$				

Symmetry code: (1) I - x, -y, -z.

All H atoms were positioned in idealized geometry and refined with individual isotropic displacement parameters using the riding model, with phenyl C-H 0.93, methylene C-H 0.97 and methyl C--H 0.96 Å.

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Data collection: XSCANS (Siemens, 1994), Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXTL/PC (Sheldrick, 1990). Software used to prepare material for publication: SHELXL93 CIFTAB.

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References

Bart, J. C. J. (1968). J. Chem. Soc. pp. 376-382.

- Jovanovski, G. & Kamenar, B. (1982). Cryst. Struct. Commun. 11, 247.
- Malik, K. M. A., Haider, S. Z., Hossain, M. A. & Hursthouse, M. B. (1984). Acta Cryst. C40, 1696-1698.
- Okaya, Y. (1969). Acta Cryst. B25, 2257-2262.
- Sheldrick, G. M. (1985). SHELXS86. Program for the Solution of Crystal Structures. University of Göttingen, Germany.
- Sheldrick, G. M. (1990). SHELXTL/PC. Version 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.
- Siemens (1994). XSCANS Users Manual. Version 2.10b. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Siemens (1996). SHELXTL/PC. Version 5. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

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Bis(µ-bicyclo[2.2.1]hept-5-ene-2-exo,3exo-dicarboxylato- $\kappa^2 O, O': \kappa^2 O'', O'''$)bis-[aqua(2,2'-bipyridine-N,N')manganese(II)] Monohydrate

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Abstract

The title compound, $[Mn(\mu-C_9H_8O_4)(C_{10}H_8N_2)(H_2O)]_2$.-H₂O, exists in the crystal as a cyclic dimer about an inversion centre. Two Mn²⁺ ions are double bridged by two bicyclic dianions which act as tetradentate ligands.

The sevenfold coordination around the Mn^{2+} ion is completed by the two bipyridine N atoms and the O atom of a water molecule. The uncoordinated water molecules are positionally disordered and connect the dimers through hydrogen bonding into chains.

Comment

The synthesis and structural characterization of 3*d*-metal complexes with bicyclo[2.2.1]hept-5-ene-2-*endo*,3-*endo*-dicarboxylic acid (*endo*-H₂L) and *N*,*N'*-donor ligands such as 2,2'-bipyridine (bpy) have been reported previously (Hartung, Baumeister, Kaplonek & Fechtel, 1993). In [Mn(*endo*-L)bpy(H₂O)₃].2.5H₂O, Mn^{II} is octahedrally coordinated by two N amine atoms, three water molecules and, surprisingly, only one carboxylic O atom of the potentially tetradentate bicyclic ligand. But the exchange of *endo*-H₂L for the isomeric 2-*exo*,3-*exo*-dicarboxylic acid (*exo*-H₂L) resulted in the title compound, (I), which has an entirely different structure from the *endo* complex.



The molecular structure of (I) and the atomic numbering scheme are shown in Fig. 1. In (I), the bicyclic ligand is indeed tetradentate, with all four O atoms participating in coordination. The complex is a centrosymmetric dimer and each Mn²⁺ ion is coordinated by two halves of the two different dianions through four O atoms, two N atoms of the amine ligand and the O atom of one water molecule. The coordination geometry is highly irregular, presumably associated with the steric constraints arising from the shapes of the polydentate ligands; perhaps the best approximation is that of a pentagonal bipyramid with N1 and O1W in the axial positions. The bpy ligand is nearly planar; the r.m.s. deviation of the atoms from the least-squares plane is 0.029 Å, indicating that the two pyridine rings do not twist significantly.

The two Mn—N bond lengths are nearly the same (see Table 1). Each of the two chelating carboxylate groups shows considerable differences between its two Mn—O distances [0.128 (2) and 0.177 (2) Å], whereas all C—O bond lengths are similar. The *exo* position



Fig. 1. Molecular structure and atomic numbering of the title compound. The displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii.

of the carboxylic groups and their *cis* conformation are evident from the torsion angles C1-C4-C5-C6-171.5(1), C2-C3-C8-C7 163.7(1) and C1-C4-C3-C2 5.0(2)°. The rigidity of the norborn-5ene skeleton is manifested by the good agreement of its geometric parameters in (I) with those observed in the above-mentioned *endo* complex (Hartung *et al.*, 1993), in bicyclo[2.2.1]hept-5-ene-2-*exo*, 3-*exo*-dicarboxylic anhydride (Ermer, Bell & Mason, 1989; Filippini, Gramaccioli, Rovere & Simonetta, 1972) and in other related compounds.



Fig. 2. Projection of the crystal packing of the title compound along [010] showing the disorder of the crystal water molecules and the hydrogen bonding (H atoms except those of the disordered crystal water molecules have been omitted for clarity).

The uncoordinated water molecule shows disorder of position about an inversion centre, the two centrosymmetrically related sites requiring an occupancy factor of 0.5. This water molecule is involved in two intermolecular hydrogen bonds bridging two dimers in this way. The complexes are thus connected into a chain parallel to the c axis. Details of the hydrogen-bonding geometry are given in Table 2 and Fig. 2 shows the crystal packing.

Experimental

The title compound, (I), was synthesized by the same procedure reported for the endo complex (Hartung et al., 1993) except for the replacement of the bicyclo[2.2.1]hept-5-ene-2endo,3-endo-dicarboxylic anhydride by its exo isomer. Suitable crystals were grown by slow evaporation from an aqueous solution at room temperature.

Crystal data

$[Mn_2(C_9H_8O_4)_2(C_{10}H_8N_2)_2-$	Mo $K\alpha$ radiation
$(H_2O)_2].H_2O$	$\lambda = 0.71073 \text{ Å}$
$M_r = 836.60$	Cell parameters from 60
Triclinic	reflections
PĪ	$\theta = 10.0 - 17.2^{\circ}$
a = 9.534(2) Å	$\mu = 0.759 \text{ mm}^{-1}$
b = 10.486(2) Å	T = 293 (2) K
c = 10.486(2) Å	Prism
$\alpha = 109.05 (1)^{\circ}$	$0.38 \times 0.28 \times 0.20$ mm
$\beta = 92.00(1)^{\circ}$	Yellow
$\gamma = 110.64 (1)^{\circ}$	
$V = 913.9(3) \text{ Å}^3$	
Z = 1	
$D_x = 1.520 \text{ Mg m}^{-3}$	
D_m not measured	

Data collection

$_{\rm ax} = 29.97^{\circ}$
$= -13 \rightarrow 13$
$= -14 \rightarrow 14$
$= -14 \rightarrow 14$
standard reflections
frequency: 60 min
intensity decay: 4.3%

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.032$ wR(F²) = 0.084 S = 1.0065315 reflections 259 parameters H atoms: see below $w = 1/[\sigma^2(F_o^2) + (0.0451P)^2]$ + 0.0847P] where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.271 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\rm min} = -0.223 \ {\rm e} \ {\rm \AA}^{-3}$ Extinction correction: none Scattering factors from International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

Mn1—O1W	2.2081 (13)	Mn103'	2.4338 (13)
Mn1-N1	2.2221 (13)	01—C1	1.256 (2)
Mn1O4	2.2573 (13)	O2—C1	1.252 (2)
Mn1—N2	2.2682 (13)	O3-C2	1.249(2)
Mn102	2.2690(12)	O4C2	1.259 (2)
Mn1—O1	2.3971 (13)	C6—C7	1.315 (3)
O1W-Mn1-N1	176.51 (5)	N1-Mn101	88.37 (4)
$O1W - Mn1 - O4^{1}$	80.22 (5)	O4 ¹ —Mn1—O1	138.42 (4)
N1-Mn1-O4'	101.92 (5)	N2-Mn1-O1	135.21 (4)
O1W-Mn1-N2	105.12 (5)	O2-Mn1-O1	55.68 (4)
N1-Mn1-N2	72.42 (5)	O1W-Mn103'	93.86 (5)
O4'-Mn1-N2	85.88 (5)	N1-Mn103'	89.63 (4)
O1W-Mn1O2	80.05 (5)	O4'-Mn1O3'	55.17 (4)
NI-MnI-O2	97.21 (5)	N2-Mn1O3'	133.07 (4)
O4'-Mn1-O2	156.07 (5)	O2-Mn1-O3'	139.79 (4)
N2-Mn1-O2	86.34 (5)	O1-Mn1-O3'	85.16 (4)
O1W—Mn1—O1	91.83 (5)		

Symmetry code: (i) -x, -y, -z.

Table 2. Hydrogen-bonding geometry (Å, °)

D — $H \cdot \cdot \cdot A$	D—H	H···A	$D \cdot \cdot \cdot A$	D — $\mathbf{H} \cdot \cdot \cdot \mathbf{A}$
$O2W - H2W1 \cdot \cdot \cdot O4'$	0.828	2.160	2.833 (4)	139
O2 <i>W</i> —H2 <i>W</i> 2···O4 ⁿ	0.850	2.084	2.871 (5)	154
Symmetry codes: (i) a	x, y, 1 + z; (i)	i) $-x, -y, -y$	- <u>z</u> .	

H-atom positions at C atoms were calculated by geometrical considerations and treated according to the riding model. All water H atoms were located by a difference Fourier synthesis and those of the disordered crystal water molecule were also treated as riding with site-occupancy factors of 0.5 on the corresponding O atoms during refinement. The positional parameters of the H atoms of the water molecule coordinated at Mn were refined.

Data collection: *DIF*4 (Stoe & Cie, 1991*a*). Cell refinement: DIF4. Data reduction: REDU4 (Stoe & Cie, 1991b). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: XP/PC (Siemens, 1990). Software used to prepare material for publication: SHELXL93.

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References

- Ermer, O., Bell, P. & Mason, S. A. (1989). Angew. Chem. 101, 1298-1301.
- Filippini, G., Gramaccioli, C. M., Rovere, C. & Simonetta, M. (1972). Acta Cryst. B28, 2869-2874.
- Hartung, H., Baumeister, U., Kaplonek, R. & Fechtel, G. (1993). Z. Anorg. Allg. Chem. 619, 1196-1202.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.
- Siemens (1990). XP/PC. Molecular Graphics Program Package for Display and Analysis of Stereochemical Data. Version 4.2 for MS-DOS. Siemens Analytical X-ray Instruments. Inc., Madison, Wisconsin, USA.
- Stoe & Cie (1991a). DIF4. Diffractometer Control Program. Version 7.08. Stoe & Cie, Darmstadt, Germany.
- Stoe & Cie (1991b). REDU4. Data Reduction Program. Version 7.03. Stoe & Cie, Darmstadt, Germany.